

Liquid drop splashing on smooth, rough and textured surfaces

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Splashing occurs when a liquid drop hits a dry solid surface at high velocity. This paper reports experimental studies of how the splash depends on the roughness and the texture of the surfaces as well as the viscosity of the liquid. For smooth surfaces, there is a “corona” splash caused by the presence of air surrounding the drop. There are several regimes that occur as the velocity and liquid viscosity are varied. There is also a “prompt” splash that depends on the roughness and texture of the surfaces. A measurement of the size distribution of the ejected droplets is sensitive to the surface roughness. For a textured surface in which pillars are arranged in a square lattice, experiment shows that the splashing has a four-fold symmetry. The splash occurs predominantly along the diagonal directions. In this geometry, two factors affect splashing the most: the pillar height and spacing between pillars.

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I. Introduction

When a liquid drop hits a solid surface, it often splashes and breaks into thousands of smaller droplets. Splashing is an excellent example of a singular breakup phenomenon with an underlying instability that is still not properly understood. As illustration, it was only recently discovered that the surrounding air pressure is an important parameter for creating a splash on a smooth dry substrate so that the splash can be completely suppressed in a low pressure environment [1]. Splashing is also broadly important in industry with applications in ink-jet printing[2], combustion of liquid fuel[3], spray drying[4] and surface coating[5].

There are two distinct types of splashing[6]: “corona” and “prompt”. Corona splashing occurs on smooth surfaces, where a symmetric corona is first formed, and droplets are ejected from the expanding corona; prompt splashing takes place on rough surfaces, where there is no corona, and droplets are created at the spreading contact line. Fig. 1 shows photographs of the two cases. A previous study proposed to explain this difference: corona splashing is caused by the effects of the air surrounding the drop and prompt splashing is caused by the effects of surface roughness [7].

Splashing has been studied since the time of Worthington in 1867 [8]. Since then, there have been many experimental studies which have attempted to find a criterion for when splashing would occur. Notably, Mundo *et al*[9] established an empirical relationship for the no-splashing to splashing transition that depended on the surface roughness, R_a , the velocity of impact, V_0 , the surface tension of the fluid, σ , the diameter of the drop, D , the dynamic viscosity of the fluid, μ , and the density of the fluid, ρ . They found that: $We^{1/2} \cdot Re^{1/4} = K_c[R_a]$,

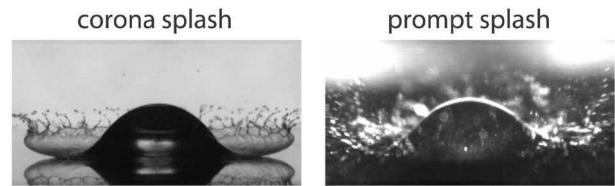


FIG. 1: Corona splash and prompt splash. The photograph on the left is a corona splash on a smooth dry surface. Droplets are created from a symmetric corona. In the photograph on the right, a prompt splash occurs on a rough dry surface. In this case there is no corona and droplets are ejected from the advancing contact line.

where We and Re are Weber number and Reynolds number respectively: $We = \rho D V_0^2 / \sigma$, $Re = \rho D V_0 / \mu$. K_c is a constant and that depends on surface roughness R_a . Splashing will occur when $K > K_c \sim 50$. When $K < K_c$ no splashing will occur. Wu[10] and Range *et al*[11] investigated the dependence of splashing on the Ohnesorge number, $Oh = \mu / \sqrt{D \sigma \rho}$. They studied the $Oh \ll 1$ case, where they could neglect the effects of viscosity, and obtained the relationship: $We_c = a \cdot \log^b R_a$, where a and b are fitting parameters. When $We > We_c$, they saw a splash. Note that neither of these relationships take into account the effects of the gas surrounding the liquid during the splash.

Some researchers also investigated the fingering instability at the rim of the expanding liquid disc. Allen[12] proposed that the Rayleigh-Taylor instability caused the fingering. Bhola *et al*[13] and Mehdizadeh *et al*[14] obtained reasonable agreement between this theory and their experiments. Thoroddsen *et al*[15] experimentally studied the fingers and proposed that the instability is caused by the presence of air trapped under the liquid drop.

This paper reports on experiments both for corona splashing on smooth dry surfaces and for prompt splashing on rough and textured dry substrates. For corona

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splashing, there are several regimes that depend on the velocity of impact and the fluid viscosity. Undulations around the rim of the spreading fluid are measured as a function of the air pressure on the smooth dry surfaces. There is a sharp jump in the number of undulations at the threshold pressure. For prompt splashing, both random roughness and roughness created by regular textured surface were studied. For a textured surface consisting of a regular array of pillars, the dependence of splashing on the vertical pillar height, lateral pillar size and pillar spacing was studied independently.

II. Corona splash on smooth surface

Previous experiment has shown that the surrounding air is crucial for corona splashing on a smooth dry surface [1]: The rows of Fig. 2 show images of splash at different background air pressures for a drop of ethanol hitting a glass substrate. Surprisingly, as the

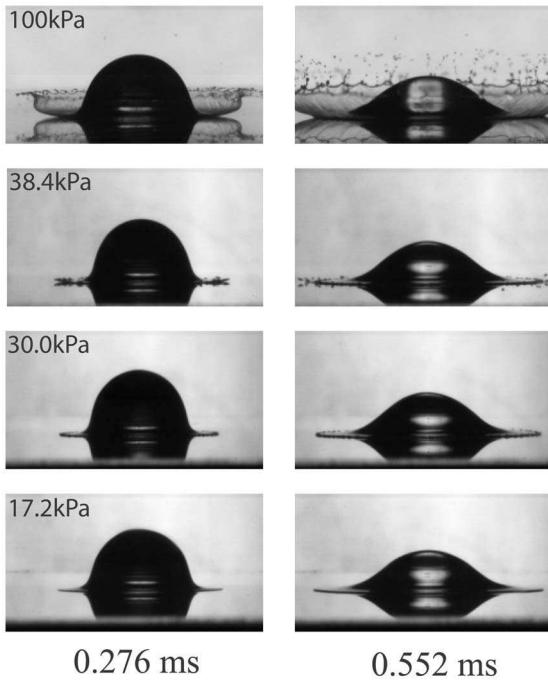


FIG. 2: Photographs of a liquid drop hitting a smooth dry substrate. A 3.4 ± 0.1 mm diameter alcohol drop hits a smooth glass substrate with impact velocity $V_0 = 3.74 \pm 0.02$ m/s at different background air pressures. Each row shows the drop at two times, 0.276 ms and 0.552 ms after impact. In the top row, $P = 100$ kPa (atmospheric pressure), the drop splashes. In the second row, at the threshold pressure, $P_T = 38.4$ kPa, the drop emits only a few droplets, traveling at a small angle with respect to the surface. In the third row, at $P = 30.0$ kPa, there is no splashing but there are undulations at the rim. In the fourth row, at $P = 17.2$ kPa, there is no splashing and no apparent undulations in the rim of the drop. Taken from [1].

pressure is lowered, fewer droplets are ejected; under low enough pressure no droplets emerge at all after impact. At a threshold pressure, P_T , the splash just begins to be formed as is shown in the second row of the figure.

The threshold pressure, P_T , as a function of impact velocity, V_0 , is shown in the main panel of Fig. 3. The curve is not monotonic. In the high velocity region above a characteristic velocity, V^* , P_T decreases as the impact velocity is raised. This is what we might naively expect. However, in the region $V_0 < V^*$, the curve is non-monotonic. This non-monotonicity indicates two different regimes at low and high velocities. Further experiments show that V^* varies with liquid viscosity and drop size [16].

Experiments have also revealed that when the surrounding gas is heavier (for example, using Kr and SF₆) and with larger liquid viscosity it is easier to create a splash. We compared two stresses[1]: the destabilizing stress from air, Σ_G , and the stabilizing stress from surface tension, Σ_L , and found:

$$\Sigma_G/\Sigma_L = \sqrt{\gamma M_G} P \sqrt{\frac{DV_0}{4k_B T}} \frac{\sqrt{\nu_L}}{\sigma} \quad (1)$$

Here γ is the adiabatic constant of the gas, M_G is the gas molecular weight, k_B is Boltzmann's constant, T is the temperature, D is the diameter of the drop, ν_L is the kinematic viscosity of liquid, and σ is the surface tension. A heavier gas or a larger liquid viscosity will increase the ratio, Σ_G/Σ_L . The ratio of these two stresses was found to be approximately constant for velocities

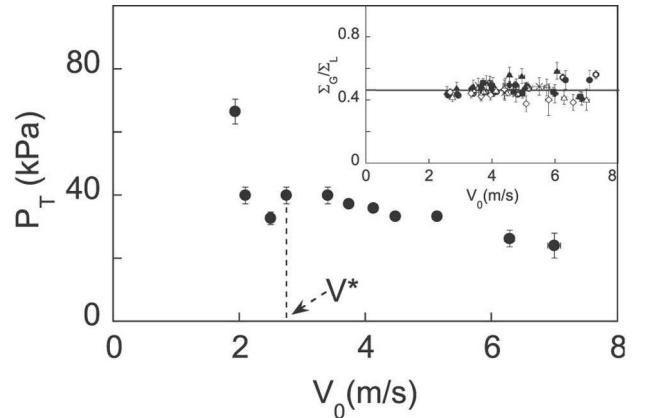


FIG. 3: Threshold pressure versus impact velocity and the collapse of data. Main panel shows P_T vs. V_0 in air. The curve is nonmonotonic: seemingly two distinct regimes are separated by a velocity, V^* . The inset plots Σ_G/Σ_L versus V_0 at threshold pressure P_T , in the region $V_0 > V^*$, for gases of different molecular weight, M_G : He($M_{He} = 4$), air($M_{air} = 29$), Kr($M_{Kr} = 84$), SF₆($M_{SF_6} = 146$) and for liquids of different viscosity, ν_L : Methanol($\nu_{Meth} = 0.68$ cSt), Ethanol($\nu_{Eth} = 1.36$ cSt), 2-Propanol($\nu_{2-Pro} = 2.60$ cSt). At threshold pressure, all Σ_G/Σ_L collapse approximately onto a constant value, 0.45. Taken from [1].

above V^* at threshold pressure. This is shown in the inset to Fig. 3 where Σ_G/Σ_L at threshold pressure P_T is plotted for gases of different molecular weights (4 ~ 146 Dalton), liquids of different viscosities (0.68 ~ 2.6 cSt) and different impact velocities (2.5 ~ 7 m/s). At threshold pressure, in the regime $V > V^*$, $\Sigma_G/\Sigma_L = 0.45$ so that Eq.1 successfully collapses all the data without any fitting parameter.

The prediction of Eq.1 that increasing the liquid viscosity leads to a lowering of the threshold pressure was verified by the data in Fig. 3 which spanned the range $0.68 \text{ cSt} < \nu_L < 2.60 \text{ cSt}$. Nevertheless this result is counterintuitive from our experience with high viscosity liquids and calls for more experiments covering a broader range of viscosity. By using silicone oils of different molecular weights, the liquid viscosity could be varied

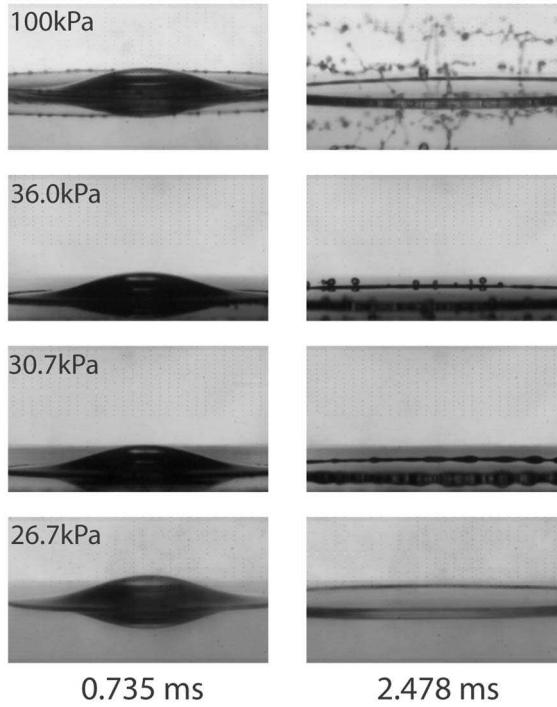


FIG. 4: Splash of a viscous drop ($\nu_L = 5 \text{ cSt}$). A 3.1 ± 0.1 mm diameter silicone oil drop hits a smooth glass substrate at impact velocity $V_0 = 4.03 \pm 0.05$ m/s under different air pressures. Each row shows the drop at two times: 0.735 ms and 2.478 ms after impact. In the top row, at $P = 100$ kPa, there is a pronounced splash. In the second row, at threshold pressure, $P_T = 36.0$ kPa, the drop just starts to splash. In the third row, at $P = 30.7$ kPa, there is no splash but there are undulations in the thickness of the rim. In the fourth row, $P = 26.7$ kPa, there is no splashing and no apparent undulations in the rim. The general property that less air leads to less splashing is similar to the low viscosity case shown in Fig. 2. However, at high viscosity splashing occurs at a later time (2.478 ms for the second row) than it does for low viscosity (0.552 ms for Fig.2 the second row).

by more than one order of magnitude, while keeping a very similar mass density ($0.82 \sim 0.95 \text{ g/cm}^3$) and surface tension ($17.4 \sim 21 \text{ mN/m}$). Fig. 4 shows photographs of a relatively viscous (5 cSt) silicone oil drop hitting a dry glass substrate under different pressures of air. Again, we find that the splash decreases as the air pressure is decreased, and that no splash occurs when the background pressure is low enough. But one difference between Fig.4 and Fig.2 is that splashing occurs at a much later time when the viscosity is large. This is most obvious if one compares the second rows. Clearly one effect of viscosity is to delay the splashing time, as we might have expected.

Fig.5 shows the threshold pressure, P_T , vs. liquid viscosity, ν_L , for 3.1 ± 0.1 mm diameter drops hitting the substrate with an impact velocity $V_0 = 4.03 \pm 0.05$ m/s. The upper curve shows, as before, the splashing threshold pressure, P_T , where splashing is first detected. The lower curve shows the threshold pressure, P_{T-bump} , for where an undulation in the expanding sheet of liquid is first observed. P_{T-bump} is defined as the lowest pressure at which undulations (or bumps) first show up, and below which no undulations can be seen. Both threshold pressures first decrease then increase with increasing viscosity. This indicates two different regimes. At low ν_L , as the viscosity is increased, the threshold pressure to create a splash decreases. Thus viscosity helps to produce splash, as predicted by Eq.1. The solid line is the

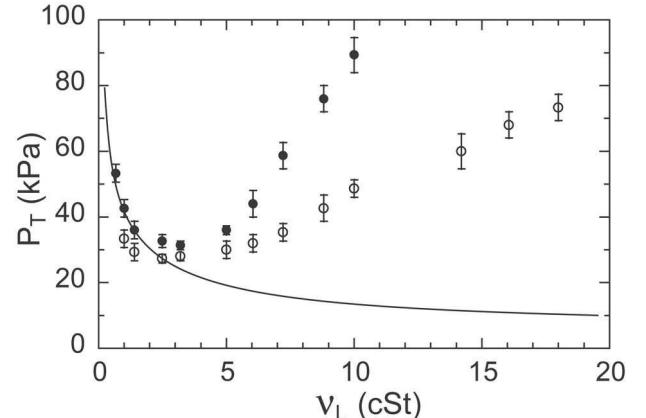


FIG. 5: Threshold pressures versus viscosity. Except for the point at the smallest viscosity, which is for methanol, the fluids were silicone oils with different molecular weights. Two curves are shown: (i) the splash threshold pressure, $P_T(\bullet)$, for a splash to appear and (ii) the bump threshold pressure, $P_{T-bump}(\circ)$, where the bump at the leading edge first appears. The curves are both non-monotonic. At low viscosity, the threshold pressures decrease with increasing ν_L whereas at high viscosity, they increase with ν_L . The solid line is the curve predicted by Eq.1: $P_T \sim 1/\sqrt{\nu_L}$. The curve fits the small viscosity regime very well, but does not capture at all the trend at high ν_L . In these experiments, the impact velocity and drop diameter are kept fixed at $V_0 = 4.03 \pm 0.05$ m/s, $D = 3.1 \pm 0.1$ mm.

scaling relation derived from Eq.1, which agrees well with the small ν_L data. However, the prediction starts to deviate at higher viscosity where the threshold pressures increase with ν_L . In this regime, the higher the viscosity, the higher the pressure of air needed to create a splash and viscosity suppresses splashing.

Why there are two different behaviors? We think that for the low ν_L regime, the expanding liquid film is stabilized mainly by surface tension so that viscosity only affects the film thickness: $d \sim \sqrt{\nu_L t}$. Thus a larger ν_L causes a thicker film which is easier to destabilize. But for the high ν_L regime, viscous drag is important and helps to stabilize the spreading drop.

III. Number of undulations versus pressure

Researchers have extensively studied the fingering instability that occurs as a splash is created [12, 13, 14, 15]. “Fingers” mean long protrusions at the rim of the expanding liquid film. In our experiment, we also observed undulations around the rim as shown in the inset to Fig.6. Here, a feature similar to “fingering” is observed. However in this case, the undulations do not extend very far out from the rim. In order to prevent possible confusion, we call them “undulations” or “bumps”. Previous studies have focused on the number of fingers as a function of impact velocity and surface roughness. Here we concentrate on the behavior and number of the bumps as the air pressure is varied.

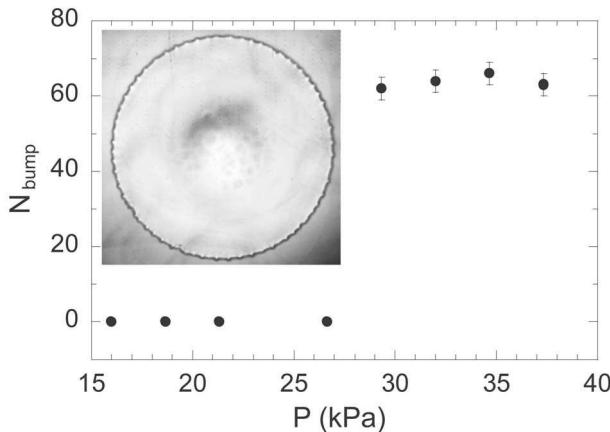


FIG. 6: Number of bumps as a function of air pressure. Inset is a bottom-view photograph showing the undulations. The number of undulations shown in the main panel, N_{bump} , is counted from such images. There is a sudden change in N_{bump} around $P_{T-bump} = 29. At that pressure N_{bump} jumps from 0 to a finite value and stays constant above that pressure.$

In Fig.6 we show N_{bump} at different pressures. We determine the number of undulations, N_{bump} , from pictures such as the one shown in the inset. At all pressures, N_{bump} is measured at the same fixed radius of expansion where the undulations are most clear. The main panel shows that at threshold pressure P_{T-bump} ,

N_{bump} jumps from zero to a finite value, and stays constant for higher pressures. The absence of undulations at low pressure suggests that no instability can grow below P_{T-bump} . The apparent pressure independence of N_{bump} above P_{T-bump} could be due to the narrow pressure range we are able to measure: above a certain pressure the entire expanding film is lifted into air.

IV. Discussion on instability mechanism

What is the mechanism for destabilizing the system and causing the occurrence of a splash on a smooth surface? What is the instability that eventually produces a splash? This question is still under debate. One prevailing theory first proposed by Allen[12], is that it is due to the Rayleigh-Taylor instability[17]. This is an instability occurring at the interface of two fluids with different densities, when the acceleration, a , points from the light fluid, ρ_1 , towards the heavy fluid, ρ_2 . Linear instability theory predicts an exponential growth in amplitude, $A \sim \text{Exp}(ct)$, with growth rate c . The wavenumber, k_m , and the growth rate, c_m , of the fastest growing mode are predicted to be[18]:

$$k_m = \sqrt{\frac{a(\rho_2 - \rho_1)}{3\sigma}} \quad (2)$$

$$c_m = \sqrt{\frac{2ak_m(\rho_2 - \rho_1)}{3(\rho_2 + \rho_1)}} \quad (3)$$

Here the light fluid is the surrounding air and the heavy fluid is the expanding liquid. The expanding disc decelerates so that the direction of a satisfies the requirement for the Rayleigh-Taylor instability. However, since $\rho_2 \gg \rho_1$, Eqs.2 and 3 do not vary appreciably with air pressure. Therefore splashing should not vary with air pressure if it were caused by Rayleigh-Taylor instability. This is inconsistent with our experiment.

Another interface instability, the Kelvin-Helmholtz instability, can take place when there is velocity jump at the interface. For inviscid fluids and $\rho_2 \gg \rho_1$, the wavenumber and the growth rate of the fastest growing mode are: [19, 20, 21]:

$$k_m = \frac{2}{3} \frac{\rho_1 u^2}{\sigma} \quad (4)$$

$$c_m = k_m u \sqrt{\frac{\rho_1}{3\rho_2}} \quad (5)$$

With u the relative velocity between two fluids at the interface. In our case, $u \sim \sqrt{DV_0/4t}$ is the velocity of the expanding liquid film. The Kelvin-Helmholtz instability strongly depends on the density of the lighter fluid, ρ_1 ,

and thus may be relevant to our experiment. However, our previous results[1] indicate that compressibility of air is important. This suggests that we should replace the Bernoulli term $\rho_1 u^2$ in Eq.4 with $\rho_1 C_G u$, with $C_G = \sqrt{\gamma k_B T / M_G}$ the speed of sound in the surrounding gas.

$$k_m = \frac{2}{3} \frac{\rho_1 C_G u}{\sigma} = \frac{2}{3} \frac{P}{\sigma} \sqrt{\frac{\gamma M_G}{k_B T}} \sqrt{\frac{DV_0}{4t}} \quad (6)$$

The characteristic length in the expanding liquid film is the film thickness, d . This suggests that the instability might be able to grow if:

$$k_m \sim 1/d \quad (7)$$

Since we also have $d \sim \sqrt{\nu_L t}$, from Eq.6 and Eq.7, we obtain as a criterion for the instability to grow:

$$\sqrt{\gamma M_G P} \sqrt{\frac{DV_0}{4k_B T}} \frac{\sqrt{\nu_L}}{\sigma} \sim 1 \quad (8)$$

The left hand side is exactly Σ_G / Σ_L in Eq.1. In the low viscosity regime, our experiment gives $\Sigma_G / \Sigma_L = 0.45$ for the splashing threshold which is consistent with the criterion in Eq.8. This suggests the possibility the Kelvin-Helmholtz instability may be the underlying instability mechanism for corona splashing.

V. Prompt splash on rough surface

A completely different type of splash, the prompt splash, occurs on rough surfaces. By systematically varying the degree of surface roughness and the air pressure, we discovered two different mechanisms for the two kinds of splashes: surrounding air is responsible for the corona splash discussed above and surface roughness is responsible for the prompt splash[7]. Under ordinary conditions (atmospheric pressure and non-zero roughness), a splash is a mixture of both contributions. By working under low pressure with a negligible amount of air, we are able to study pure prompt splashing.

Since a prompt splash is caused by surface roughness, it may retain information about surface roughness in the distribution of sizes of the ejected droplets. We find that this is the case. We mix a small amount of ink into our ethanol and then collect the ejected droplets on a sheet of white paper. We then obtained the sizes of ejected droplets by measuring the size and darkness of the stains left on the paper. Our previous study[7] shows that the number of droplets, N , decays exponentially with their radius, r : $N(r) \sim \text{Exp}(-r/r_0)$ (see inset of Fig. 9(b)). This indicates the existence of a characteristic decay length, r_0 .

The decay length, r_0 , correlates with roughness of the surface, R_a [7]. At small R_a , we have the relationship $r_0 \approx R_a$; but for large roughness, this breaks down as r_0 saturates at a constant value. We can understand

this behavior in the following manner. After impact, the thickness of expanding film, d , grows continuously from being molecularly thin just after the impact, to approximately $40 \sim 50 \mu\text{m}$ at the end of the film expansion. When the surface roughness is small, d can grow to be much larger than R_a . At the beginning, when d is small, the film is thinner than R_a and continues to eject droplets until the film becomes much thicker than the surface roughness. After d grows to be larger than R_a , the roughness is too small to destabilize the liquid film and produce a splash. Thus the distribution of ejected droplets reflects the surface roughness R_a and we find $r_0 \approx R_a$. However, when the roughness is large, d can never grow to be greater than R_a . Consequently, r_0 can only grow to the maximum size of d at its final thickness. This is consistent with the decay constant, r_0 , saturating around $40 \mu\text{m}$, which is roughly the film thickness at the end of expansion.

VI. Prompt splash on textured surface

The last section showed that roughness has a strong effect on prompt splash. To understand this dependence in more detail and to understand how surface properties affect splashing, we study splashing

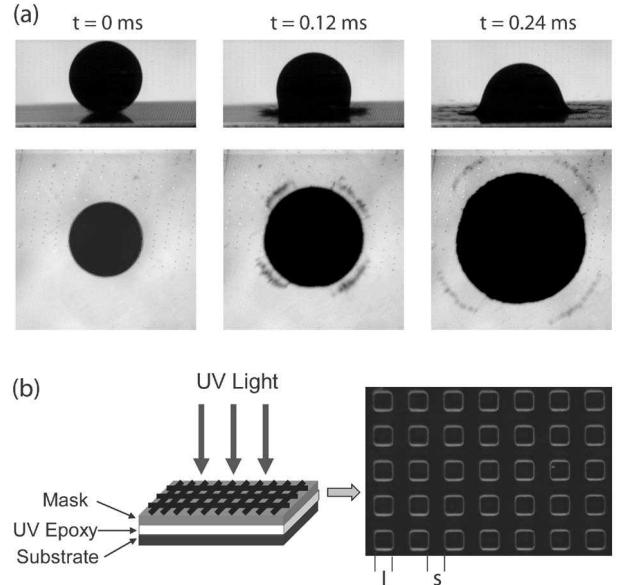


FIG. 7: Prompt splashing on a textured surface. (a) Top row shows a side view of prompt splashing on a textured surface. Bottom row shows a bottom view of prompt splashing on a textured surface. There is a clear four-fold symmetry in the splash which is predominantly in the diagonal directions of the square lattice created by the pillars. (b) Making a textured surface with UV-lithography technique. Left cartoon shows the UV-lithography process. The right picture shows a typical textured surface under the microscope. We define the pillar height as h , lateral pillar size as l and spacing between pillars as s . For this particular substrate, $h = 18 \mu\text{m}$, $l = s = 60 \mu\text{m}$.

on a well-defined textured surface of regular patterns.

The textured surface is made with UV-lithography: We first spin coat UV epoxy (SU8-2000, MicroChem Corp.) onto a clean glass microscope slide. We then cover the slide with a mask of pre-designed pattern (square blocks in a square lattice) and expose the slide to UV light. After development, the UV epoxy film which is directly under the transparent part of the mask will harden and the rest of it can be rinsed away, resulting in a structure on the substrate of square pillars arranged in a 2D square lattice as shown in Fig. 7b. There are three important quantities relating to our splash experiments in this textured surface: (1) the vertical pillar height, h , (2) the lateral pillar size, l , and (3) the lateral spacing between pillars, s . By changing the spin speed, we can vary pillar height, h ; by designing different mask patterns, we can vary both l and s independently. Thus we can vary every aspect of the structure.

Fig. 7(a) shows photographs of a prompt splash on a textured surface under low air pressure. The impact velocity is $4.3 \pm 0.1 \text{ m/s}$, and the drop diameter is $D = 3.4 \pm 0.1 \text{ mm}$. The top row shows a side view of the splash. It has a similar look as the prompt splash on an ordinary rough surface. However, the bottom view shown in the second row reveals a very striking feature: the splashing occurs with four fold symmetry. The droplets are ejected predominantly along the diagonal directions of the square lattice. Fig. 7(b) shows the process of UV-lithography and a picture of textured surface under a microscope.

We can now vary the profiles of the surface and determine their effect on the splash and the ejected droplet distribution. Again we use the ink spot technique to measure the size distribution of droplets, as mentioned in last section. In the first set of experiments we first keep lateral size constant at $l = s = 60 \mu\text{m}$, and vary the vertical height of the pillars, h . Fig. 8(a) shows the number of droplets, N , versus their radius, r . Similar to the case with random roughness, we find an exponential decay at large r , $N(r) \sim \exp(-r/r_0)$, with a characteristic decay length, r_0 , that varies with h . Fig. 8(b) shows that r_0 varies with h in a nonmonotonic manner. For small h , r_0 increases with, and has a value comparable to, h . This indicates that r_0 is determined by h . However, when h is greater than $18 \mu\text{m}$, the opposite trend occurs: r_0 decreases as h increases. Fig. 8(c) shows the sum of the areas created by all of the ink spots, A_{tot} , as a function of h . A_{tot} is a quantity that indicates the total amount of ejected droplets. Fig. 8(b) and (c) have the same shape, indicating that r_0 and A_{tot} are strongly correlated. The decreasing trend for large h implies that larger roughness leads to less splashing. When h is greater than $60 \mu\text{m}$, there is no splash at all.

These results are counterintuitive. We suspect they are caused by the way in which the impacting liquid drop can flow between the channels set up by the pillar structure. At small pillar heights, the liquid in the drop can easily reach the bottom of the canyon between the pillars and can then expand along the bottom surface. During

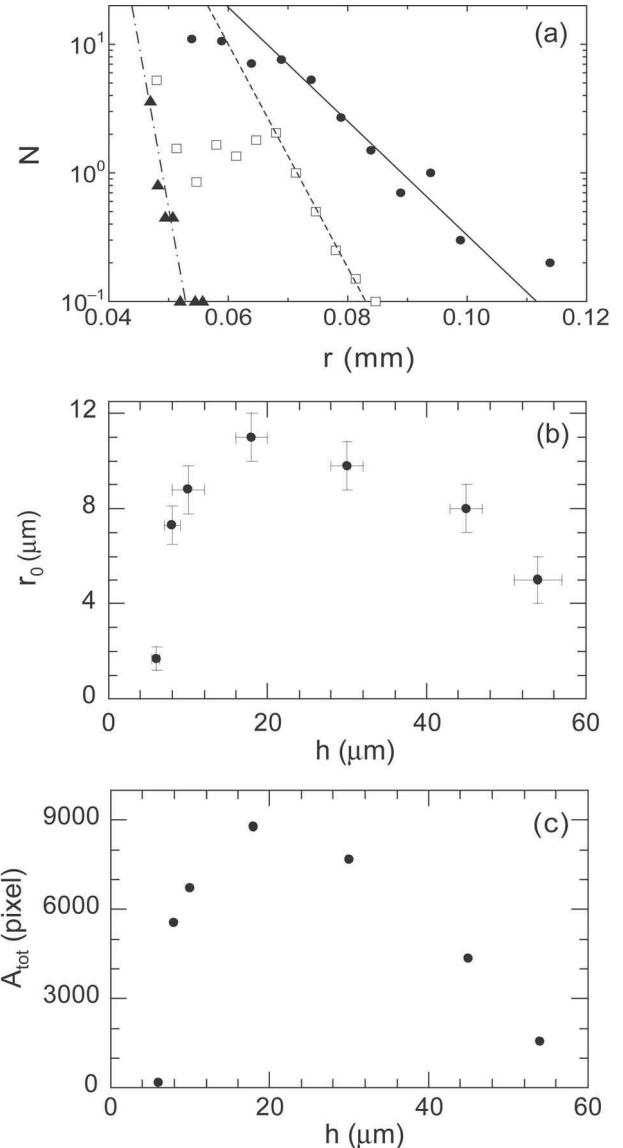


FIG. 8: Decay length and total splash area versus pillar height. (a) The number of ejected droplets, N , as a function of droplet radius, r , for three pillar heights, h . The lateral dimensions are kept fixed at $l = s = 60 \mu\text{m}$. The straight lines are an exponential fit to the large r tail of the distribution: $N \sim \exp(-r/r_0)$, with $r_0 = 0.0017 \text{ mm}$ (---), 0.0098 mm (—) and 0.005 mm (---) for $h = 6 \mu\text{m}$ (▲), $h = 30 \mu\text{m}$ (●), and $h = 54 \mu\text{m}$ (□) respectively. Note that r_0 does not vary monotonically with h . (b) The exponential decay length, r_0 , versus pillar height, h . r_0 first increases then decreases as h increases. (c) The total ink spot area, A_{tot} , as a function of h , for the same set of experiments shown in (b). The curve in (c) has a similar shape as (b) indicating that A_{tot} and r_0 are strongly correlated.

expansion, the liquid film is destabilized by the pillars, producing droplets with a size related to the pillar height, h . This produces a positive correlation between r_0 and h for $h < 18 \mu\text{m}$. However, as h increases, it is increas-

ingly difficult for the impacting drop to reach the bottom surface. In this case, we suspect that much of the drop expands on top of the pillars, rather than between them. Once $h \approx 60\mu\text{m}$, that is when the height is about the same size as the lateral dimensions l and s , the situation resembles a drop expanding on a flat surface with many holes rather than pillars. Here all the obstructions are underneath the liquid film and make only a small perturbation to its expansion. Because the impacting drop can only penetrate a finite depth below the pillar top, it does not know how far away it is from the bottom surface. This suggests that the amount of splashing should saturate as the pillar height is increased. We do not have a good explanation of the surprising fact that the splashing can be completely eliminated if the pillars are sufficiently tall.

To some extent, this is similar to the Cassie state of a drop on a superhydrophobic rough surface studied by Quéré *et al*[22, 23], where a water drop can sit on top of air trapped in the rough profile of the substrate. However we note that their case is static whereas ours is probably driven by the fast dynamics of the expanding drop. Moreover in the case studied by Quéré *et al* the air plays an important role in supporting the weight of the drop. In our situation air has been pumped out of the system.

To understand the effect of the lateral dimension on the splashing, we make substrates of different l and s , while keeping h fixed. Fig. 9(a) main panel shows N versus r with an exponential fitting function, $N \sim \text{Exp}(-r/r_0)$, for different lateral sizes. Fig. 9(b) plots r_0 as a function of lateral pillar size l and spacing s . As l and s are varied we keep $l = s$, and $h = 10\mu\text{m}$. Fig. 9(c) shows the total area, A_{tot} , vs. l and s . A_{tot} has the same dependence on l and s as does r_0 . Both quantities increase with lateral size in most of our range, then decrease at the end. This means that increasing the lateral dimensions will enhance splashing for small l and s . When the pillars are too sparse, splashing becomes less pronounced suggesting that it is more difficult to destabilize the liquid film. We should also note that r_0 is always much smaller than l and s , while much closer to the pillar height $h = 10\mu\text{m}$. This indicates that h is more important in determining r_0 than are l and s .

A comparison of a textured surface with a random roughness surface is shown in the inset of Fig.9(a). Both curves decay exponentially, but random roughness curve (upper curve) has a much larger r_0 . This is surprising because both curves have similar roughness ($l = s = 20\mu\text{m}, h = 10\mu\text{m}$ for the textured surface and $R_a = 16\mu\text{m}$ for the case of random roughness). We can understand this qualitatively using the data shown in Fig.8(b). There we see r_0 starts to decrease at $h \approx \frac{1}{3}l$. Thus here at $h = 10\mu\text{m} = \frac{1}{2}l$, it is already difficult for the drop to reach the bottom of the substrate and get destabilized. However, the random roughness is made by particles coated on surface. This substrate never resembles a flat surface with many holes. Therefore the random roughness surface makes a much larger splash

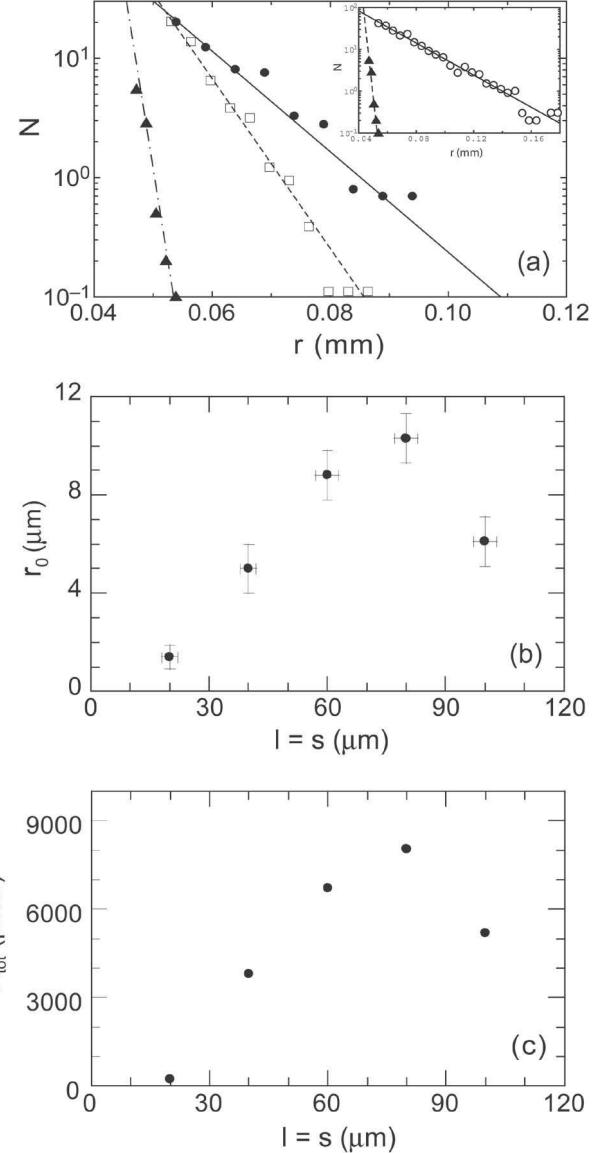


FIG. 9: Decay length and total splash area versus lateral size of the pillars. (a) N as a function of r for three lateral sizes. The pillar height h is fixed at $h = 10\mu\text{m}$. The straight lines are an exponential fit to the large r tail of the distribution: $N \sim \text{Exp}(-r/r_0)$, with $r_0 = 0.0014\text{mm}$ (---), 0.010mm (—) and 0.0061mm (---) for $l = s = 20\mu\text{m}$ (▲), $l = s = 80\mu\text{m}$ (●), and $l = s = 100\mu\text{m}$ (□) respectively. Note that r_0 does not vary monotonically with l and s . Inset compares the distribution for a textured surface with $l = s = 20\mu\text{m}, h = 10\mu\text{m}$ (▲) with a sample with comparable random roughness $R_a = 16\mu\text{m}$ (○). The random roughness creates a much larger splash. (b) r_0 is plotted versus l and s . r_0 first increases then decreases with l and s . (c) A_{tot} versus l and s . Again we see a similar shape as in (b).

with a larger r_0 .

Fig. 9 demonstrates that r_0 changes as we vary l and s together. One further question is whether this is caused by a change in l or a change in s or in both? We can check

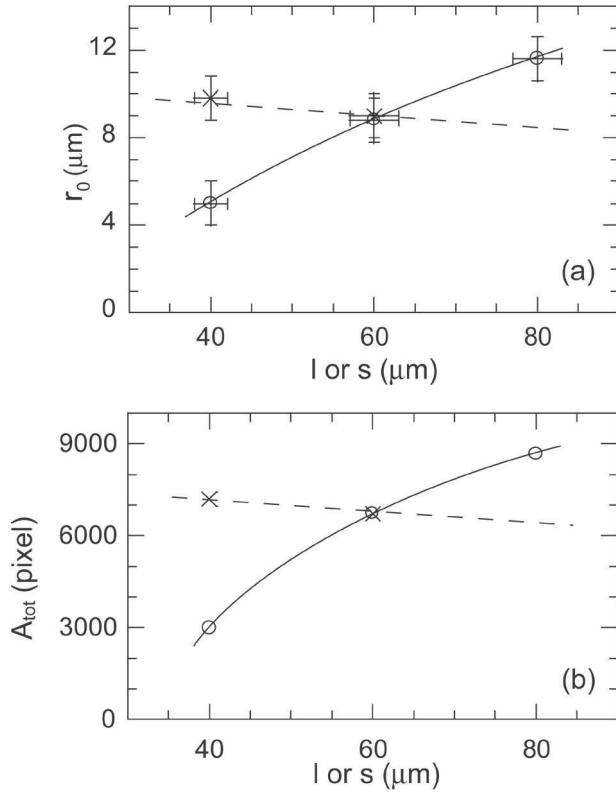


FIG. 10: Effect on splash characteristics of varying l as compared to the effect of varying s . (a) We independently vary lateral pillar size , $l(\times)$, and spacing between pillars, $s(\circ)$, to compare their effect. Two lines are guides to the eye for l (---) and s (—). Pillar height is kept fixed at $h = 10\mu\text{m}$. When l is varied, $s = 60\mu\text{m}$ is held fixed; when s is varied, $l = 60\mu\text{m}$ is held fixed. The comparison shows that r_0 changes with s but not l . (b) A plot of the total area, A_{tot} , as a function of l or s , for the same experiment as in (a). It produces the same trend as in (a).

this by varying l and s independently. Fig. 10 shows the result. The different symbols are for varying l and s separately while leaving all other conditions unchanged. Apparently increasing the spacing between pillars, s , enhances r_0 and A_{tot} while l has a much smaller effect on the splashing behavior.

The fact that increasing s enhances splashing helps to explain why we see splashing in the diagonal directions in Fig. 7a. Along the diagonal, the distance between pillars is the greatest. Because s is largest in those directions, splashing preferentially occurs in those directions.

The textured substrate not only affects the prompt splashing caused by surface roughness, it also changes the behavior of corona splashing caused by the surrounding air. Fig. 11(a) shows a typical corona splash on a smooth surface at atmospheric pressure, while Fig. 11(b) shows, at the same pressure, no splash at all on a textured surface consisting of tall pillars. In both experiments the drop hits the substrate at the same impact velocity. Different amounts of splashing can also be achieved by

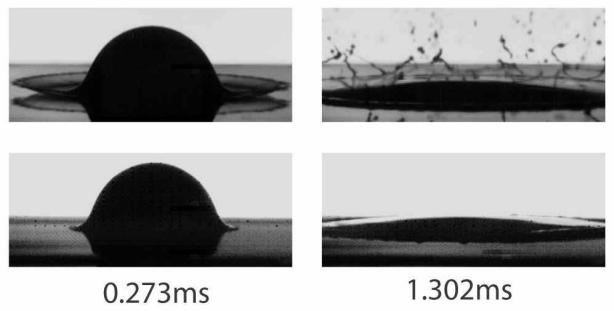


FIG. 11: Effect of pillars on suppressing corona splashing. All experiments are done at atmospheric pressure, with $V_0 = 4.3\text{m/s}$ and $D = 3.4\text{mm}$. (a) A corona splash on a smooth surface. (b) Splashing is completely suppressed on a surface consisting of pillars with $l = s = 60\mu\text{m}$ and $h = 125\mu\text{m}$. We can tune the amount of splashing by varying the pillar height even under atmospheric pressure.

creating pillars with intermediate heights. These results suggest that the pillars form channels through which the air can escape so that the importance of the air for creating the splash is minimized. This discovery demonstrates another way in which one can suppress splashing. Moreover, it has the advantage that this suppression can be achieved without decreasing the gas pressure.

VII. Conclusion

This paper systematically studied the splashing of liquid drops on various dry solid surfaces. This study corroborates that there are two mechanisms corresponding to the two kinds of splashes. Air causes the corona splash on smooth dry surfaces and substrate roughness causes the prompt splash. For the corona splash, we discovered several regimes. At high impact velocity, there are two regimes as the viscosity of the liquid is varied. We also studied the finger instability as function of air pressure and find a jump in the number of bumps. We suspect that Kelvin-Helmholtz instability coupled with the compressibility of air, is a possible mechanism for the splashing instability. This mechanism agrees well with our experimental data.

In order to examine the effect of surface roughness, we studied splashing on textured surfaces consisting of square pillars arranged in a square lattice. We found that the dimensions of the pillars strongly affect splashing. Here the pillar height, h , is found to be the most important factor determining characteristic decay length, r_0 . We discovered that the splash preserves the symmetry of substrate. This shows that the splash direction can be controlled. We also find that corona splash under atmospheric pressure can be suppressed by making tall pillars on surface. This provide another way to reduce splash even under normal pressure. Since splashing is involved in many industrial processes[2, 3, 4, 5], these discoveries could have important practical applications.

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